

# The Isotope Geochemistry of N<sub>2</sub> in Natural Gas Pools

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**Abstract** The isotopic compositions of N<sub>2</sub> (  $^{15}\text{N}$ ,‰, ATM) imply the geochemical origins of molecular nitrogen in gas pools. N<sub>2</sub> with  $-19\text{‰} < ^{15}\text{N}_{\text{N}_2} < -10\text{‰}$  may be from immature sedimentary organic matter.  $-10\text{‰} < ^{15}\text{N}_{\text{N}_2} < -2\text{‰}$  may indicate the N<sub>2</sub> originates from mature (including high mature) sedimentary organic matter.  $-2\text{‰} < ^{15}\text{N}_{\text{N}_2} < +1\text{‰}$  implies the N<sub>2</sub> may be from the deep crust or mantle. N<sub>2</sub> with  $^{15}\text{N}_{\text{N}_2} = 0\text{‰}$  and N<sub>2</sub>/Ar=38 ~ 84 suggests to come from the atmosphere.  $-1\text{‰} < ^{15}\text{N}_{\text{N}_2} < +4\text{‰}$  characterizes N<sub>2</sub> from ammonium clay minerals in shale and mudrock during metamorphism.  $^{15}\text{N}_{\text{N}_2} = +4\text{‰}$  is the typical feature of N<sub>2</sub> from saltpeter in evaporite.  $+4\text{‰} < ^{15}\text{N}_{\text{N}_2} < +18\text{‰}$  indicates N<sub>2</sub> may derive from post-mature sedimentary organic matter, which is the main source of N<sub>2</sub>-high gas (N<sub>2</sub> > 60%) pools. N<sub>2</sub>-high gas pools, which occurs in a large scale in periferous basin, suggests that it is mainly the post-mature gas trapped in the reservoir. N<sub>2</sub> with  $^{15}\text{N}_{\text{N}_2}$  varying from  $-10\text{‰}$  to  $-2\text{‰}$  is the main source of N<sub>2</sub>-rich gas (N<sub>2</sub> > 15%) pools, and with  $^{15}\text{N}_{\text{N}_2}$  varying from  $+1\text{‰}$  to  $+4\text{‰}$  may form either the N<sub>2</sub>-high or the N<sub>2</sub>-rich gas pools.

**Key words** nitrogen isotope origin of nitrogen natural gas ammonia sedimentary organic matter

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## INTRODUCTION

gas pools (Zhu,1994).N<sub>2</sub>-rich gas pools have been discovered in many petroliferous basins all over the world (Headlee,1962),such as Mid-European basin(Littke et al.,1995;Krooss et al.,1995),Volga-Ural basin(Prasolov et al. ,1990),Zhungaer basin (Wang and Jiang,1997), Yinggehai basin, Sanshui basin (Zeng ,1986),Subei basin (Xu,1994),Songliao basin (Huang,1995),Great Valley basin (Jenden et al.,1988a) and Western Interior basin (Hoering and Moore,1958;Jenden et al.,1988b).

However, the isotope geochemistry of N<sub>2</sub> in gas pools is still poorly known (Krooss et al.,1979;Prasolov et al.,1990;Xu,1994;Witicar,1994;Littke et al.,1995;Krooss et al.,1995; Du et al.,1996). We believe that this is due to multiple origins which are often mixed together and lead to irregular variations of <sup>15</sup>N of N<sub>2</sub> in gas pools. In order to unravel the molecular nitrogen isotopic compositions in gas pools,the isotope geochemistry of single and typical sources of N<sub>2</sub> has to be studied.

In order to elucidate the geochemical implications of isotopic compositions of molecular nitrogen in gas pools,we chose four typical basins(i.e. the West-Siberian basin in Russia,the Yinggehai basin of China,the Californian Great Valley basin of USA and the Mid-European basin) among the numerous N<sub>2</sub>-bearing basins in the world.The four basins have significantly different isotopic characteristics of gases,which may provide insight into the isotope geochemistry of N<sub>2</sub> in natural gas pools.

## EXPERIMENTAL AND RESULTS

Thirty five gas samples were collected at high pressure in steel bottles with stopcocks at both ends from the gas-producing wells in Yinggehai basin ,offshore and southwest of Hainan Island, China. In the laboratory,chemical compositions of major,minor and trace components (CH<sub>4</sub>、 C<sub>2+</sub>、 CO<sub>2</sub>、 N<sub>2</sub>、 Ar and He) were analyzed using a gas chromatograph system.Peak heights of chemical components in the samples were calibrated against those of the in-house standard gases.Overall analytical errors were estimated to be <5% for most samples.

Purification of He and Ar was carried out by a two-stage Ti-Zr getter in a special high vacuum stainless steel system. An activated charcoal trap was held at liquid N<sub>2</sub> temperature in order to separate Ar from He. A VG-5400 static mass spectrometer was used for the measurements of <sup>3</sup>He/<sup>4</sup>He and <sup>40</sup>Ar/<sup>36</sup>Ar ratios. The precision of the measurement was less than 1% for most samples.

Separation of the gas samples into pure gas components(CH<sub>4</sub>、 CO<sub>2</sub> and N<sub>2</sub>) and oxidation of CH<sub>4</sub> to CO<sub>2</sub> were performed in a glass vacuum line using a trap held at liquid N<sub>2</sub> temperature,another trap at acetone-liquid N<sub>2</sub> temperature,and a CuO furnace.

The  $^{13}\text{C}$  and  $^{15}\text{N}$  were determined by a MAT-251 mass spectrometer(Finnigan). The respectively.

## FOUR TYPICAL $\text{N}_2$ -BEARING BASINS

### West-Siberian basin

The West-Siberian province is a great gas-bearing basin formed in Mesozoic time. More than 20 giant and super-giant gas fields have been discovered in the northern part of the basin. Both the two sets of the source rocks and reservoirs are from the Cretaceous (Dai, 1985; Zhang, 1990).

The Cenomanian reservoirs contain a huge gas accumulation of either bacterial or early thermogenic origin. The source rock is a coal-bearing series of Lower Cenomanian, Albian and Aptian age with immature organic matter ( $R_o < 0.6\%$ ). Sandstones in the Upper Cenomanian are the principal gas reservoirs, which produce dry gas ( $C_1/C_{1-2} > 0.99$ ,  $\text{CH}_4 = 95.5\% \sim 99.7\%$ ,  $\text{C}_2\text{H}_6 = 0.02\% \sim 0.08\%$ ,  $\text{C}_3\text{H}_8 = 0.03\% \sim 0.025\%$ ).  $\text{N}_2$  and  $\text{CO}_2$  also exist commonly in these gas reservoirs, which the volume content of  $\text{N}_2$  ( $0.1\% \sim 1.2\%$ ) being higher than that of  $\text{CO}_2$  ( $0.1\% \sim 0.3\%$ ). The carbon isotope values of methane are light ( $^{13}\text{C} = -68\% \sim -55\%$ , PDB) (Dai, 1985) and the nitrogen isotope values of  $\text{N}_2$  are also light ( $^{15}\text{N} = -19.0\% \sim -10.7\%$ ) with a variability from  $-16\%$  to  $-12\%$ . The associated  $^3\text{He}/^4\text{He}$  ratios are at  $10^{-8}$  order of magnitude (Prasolov et al., 1990).

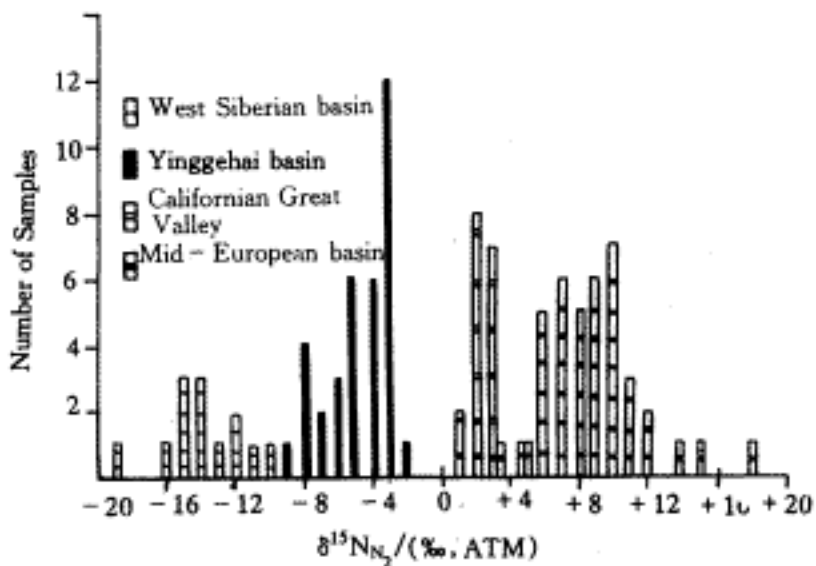


Fig.1 Histograms of  $^{15}\text{N}_{\text{N}_2}$  of natural gases from the West Siberian basin (Prasolov et al.,1990), Yinggehai basin, Californian Great Valley (Jenden et al., 1988a) and Mid-European basin (Boigk et al., 1976; Weinlich, 1991)

### Yinggehai basin

Yinggehai basin is a gas-rich basin formed in Cenozoic times. Three giant gas fields have been discovered in its southern part. The source rock is a neritic and bathyal dark mudstone in the Pliocene Yinggehai formation with mature (mainly high mature) humic organic matter. According to the stratigraphy of the Qiongdongnan basin adjacent to Yinggehai basin, other potential source rocks are marine dark mudstones of the Miocene Meishan formation and Oligocene limnic mudstones and coal-bearing series.

The detected natural gas in Yinggehai basin is primarily produced from marine sandstones of the Pliocene Yinggehai formation. Hydrocarbon composition in natural gas varies in a wide range, the contents of  $\text{CH}_4$ ,  $\text{C}_2\text{H}_6$  and  $\text{C}_3\text{H}_8$  varying from 6.47% to 90.99%, 0.013% to 2.09%, and 0.002% to 0.55%, respectively, and  $\text{C}_1/\text{C}_{1-5}$  ratios varying from 0.96 to 1.00.  $\text{CO}_2$  (0.10% ~ 88.93%) and  $\text{N}_2$  (2.93% ~ 33.48%) are common in the natural gas, which appear to be inversely correlated to each other. Methane  $^{13}\text{C}$  values vary from -63.14‰ to -29.08‰, but usually from -40‰ to -30‰, and nitrogen  $^{15}\text{N}$  values vary mainly from -9‰ to -2‰.

### Californian Great Valley basin

The Californian Great Valley basin is rich in oil and gas, which was formed in the late Mesozoic times. It is separated into two main basins, the Sacramento to the north and the San Joaquin to the south by Stockton arch. The northern San Joaquin basin and the Sacramento basin are dry gas regions, where many gas fields have been discovered (Jenden et al., 1988a). Franciscan metasedimentary complex is the basement of this basin and Cretaceous and Tertiary siltstones, shales and sandstones were deposited above. Gas reservoirs are primarily from the Cretaceous, Miocene and Pliocene (Jenden et al., 1988a).

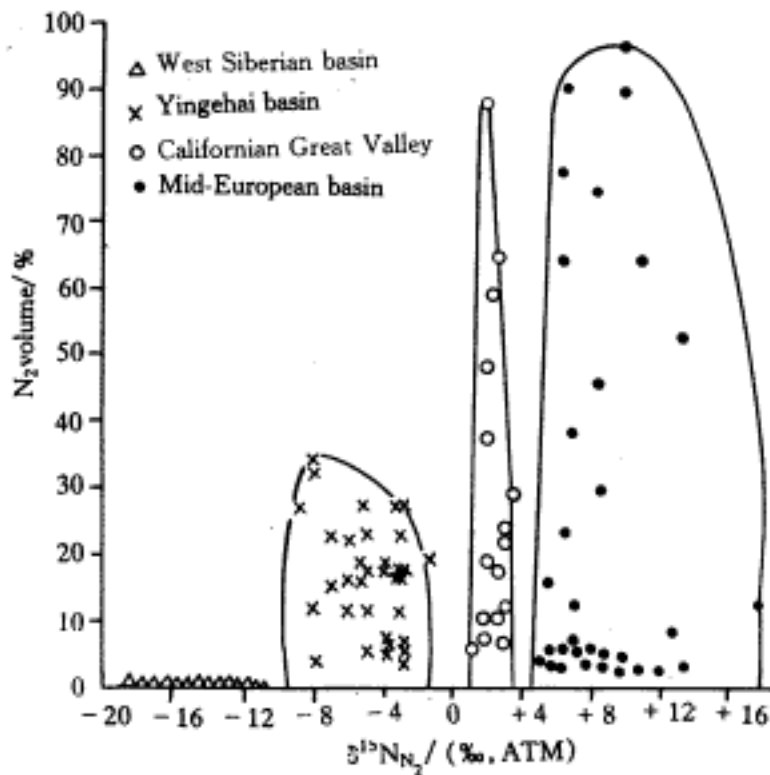


Fig.2 Plot of  $N_2$  versus  $^{15}N_{N_2}$  of natural gases from the West Siberian basin (Prasolov et al.,1990),Yinggehai basin,Californian Great Valley (Jenden et al.,1988a) and Mid-European basin (Boigk et al.,1976;Weinlich,1991)

The concentrations of methane in natural gas vary from 12% to 96%,  $C_{2+}$  ~ 0.02% to 5.1%,  $CO_2$  ~ 0.001% to 0.6%, and  $N_2$  ~ 1.0% to 87%. Methane  $^{13}C$  values vary from -52.4‰ to -15.2‰, and nitrogen  $^{15}N$  values vary from +0.9‰ to +3.5‰ with a main range from +1.0‰ to 3.0‰ (Jenden et al.,1988a).

### Mid-European basin

The Mid-European basin is a gas-rich basin formed in late Paleozoic to Cenozoic times. Lots of  $N_2$ -rich gas fields have been discovered in its northern part. The source rocks are coal-bearing series in the Upper Carboniferous with post-mature organic matter. Gas reservoirs are primarily in Permian Rotliegend and Triassic Buntsandstein formations. Main components of natural gas are  $CH_4$  and  $N_2$ . The content of  $N_2$  is more often greater than 50% in the gas accumulations in the northern part of the basin. The gas occurrence with  $N_2$  higher than 50% is consistent with the high maturity of the source rocks with vitrinite reflectance values higher than 3.0% (Littke et al.,1995; Krooss et al.,1995). The content of  $CO_2$  is less than 1%. Methane  $^{13}C$  values are greater than -36‰, and can be as high as -15‰. Molecular nitrogen  $^{15}N$  values vary from +6.5‰ to +18.0‰ (Maksimov et al.,1975; Stahl,1977).

## DISCUSSION

Heering and Moore(1958) discovered the  $^{15}\text{N}$  values of  $\text{N}_2$  in Oklahoma, Mississippi and Texas to vary from  $-10.5\text{‰}$  to  $-2.1\text{‰}$ , but to vary from  $+1.3\text{‰}$  to  $+11.9\text{‰}$  in Arkansas. They gave no interpretation for the geochemical implications of the two groups of molecular nitrogen  $^{15}\text{N}$  values. Bokhoven and Theeuween(1966) measured the molecular nitrogen  $^{15}\text{N}$  values in Dutch gas reservoirs, but they could not affirm the origin of molecular nitrogen. Maksimov et al. (1975) studied the molecular nitrogen  $^{15}\text{N}$  values in  $\text{N}_2$ -high gas reservoirs of Mid-European basin, which vary from  $+4.28\text{‰}$  to  $+18\text{‰}$ , and supposed the origin of molecular nitrogen should be derived from deep crystalline rocks. But this inference was denied by Littke et al.(1995) and Krooss, et al.(1995). Jenden et al.(1988a) thought the molecular nitrogen with  $^{15}\text{N}$  values varying from  $+0.9\text{‰}$  to  $+3.5\text{‰}$  in gas reservoirs of the California Great Valley originated from the upper Jurassic Franciscan metasedimentary complex, according to studies on the isotope ratios of carbon, nitrogen and helium. Prasolov et al.(1990) discovered that the molecular nitrogen  $^{15}\text{N}$  values in the former USSR could be divided into two groups. One of them has  $^{15}\text{N}$  values of about  $-10\text{‰}$ , the other about  $+14\text{‰}$ . The "light" nitrogen originates from organic matter and presumably from ammonia salts via the gaseous ammonia stage.

Fig.1 shows the features of isotopic compositions of the molecular nitrogen in the four typical  $\text{N}_2$ -bearing basins. It appears that the molecular nitrogen  $^{15}\text{N}$  values differentiate among those basins and have multiple central distribution patterns different from the two group patterns suggested by Prasolov et al.(1990). Fig.2,3, and 4 show the plots of volume content of molecular nitrogen vs molecular nitrogen  $^{15}\text{N}$  values, the molecular nitrogen  $^{15}\text{N}$  values vs the associate methane  $^{13}\text{C}$  values, and  $\text{N}_2/\text{Ar}$  ratios vs the molecular nitrogen  $^{15}\text{N}$  values, respectively, which also reflect multiple central distribution patterns.

In order to understand the implications of the multiple central distribution patterns of molecular nitrogen  $^{15}\text{N}$  values in petroliferous basins, it is necessary to introduce the possible origins of molecular nitrogen in subsurface. The possible nitrogen sources include: (1) atmosphere, (2) immature sedimentary organic matter, (3) mature (including high-mature) sedimentary organic matter, (4) post-mature sedimentary organic matter, (5) ammonium clay minerals in sedimentary rocks, (6) radiogenic nuclear reaction, (7) nitrogen-bearing salt in evaporite; (8) magmatic rocks, and (9) the deep crust or mantle degassing.

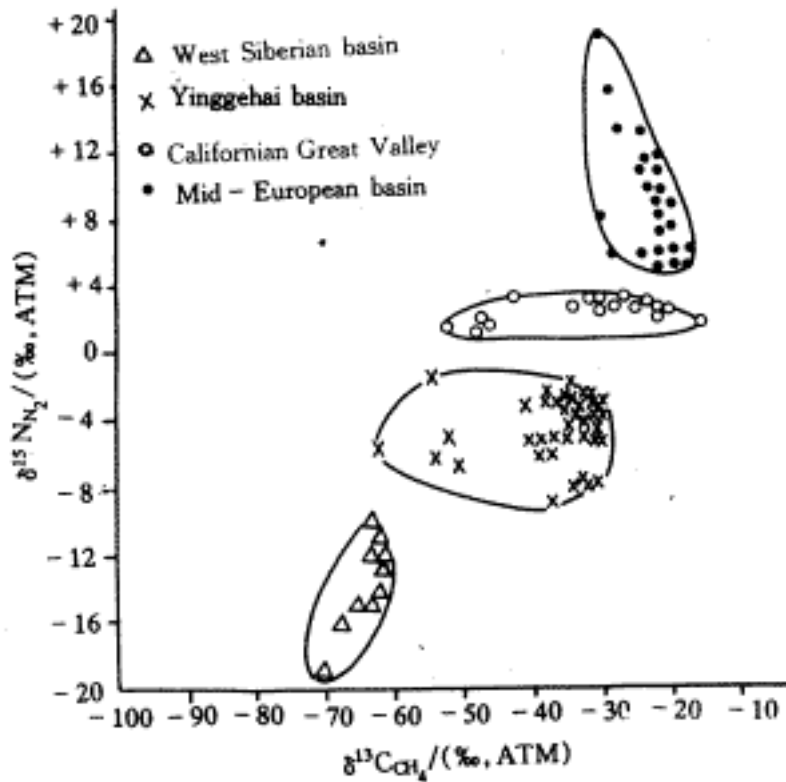


Fig.3 Plot of  $^{15}\text{N}_{\text{N}_2}$  versus  $^{13}\text{C}_{\text{CH}_4}$  of natural gases from the West Siberian basin (Dai,1985;Zh-ang,1990;Pra solov et al.,1990), Yinggehai basin,Californian Great Valley (Jenden et al.,1 988a),and Mid-European basin(Boigk et al.,1976,Stahl 1977;Weinlich,1991)

The molecular nitrogen from the atmosphere is characterized by the  $+15\text{N}$  values = 0‰ and  $\text{N}_2/\text{Ar}$  ratio = 38 ~ 84. This feature is considered to be unchanged in the past 3 billion years (Sano and Pillinger, 1990). The atmospheric  $\text{N}_2$  is not the main source for  $\text{N}_2$  in natural gas accumulations, which the evidence is that the isotopic compositions of  $\text{N}_2$  in  $\text{N}_2$ -rich gas reservoirs over the world have no signs of atmospheric characteristics (Heoring and Moore, 1958; Bokhoven and Theeuween, 1966; Maksimov et al., 1975; Prasolov et al., 1990; Littke et al., 1995; Krooss et al., 1995). The radiogenic molecular nitrogen can be neglected because of its little contribution to gas pools, and magmatic rocks-derived  $\text{N}_2$  also has little significance for it (Krooss et al., 1995) because nitrogen content in magmatite is the lowest in all crust rocks (Littke et al., 1995). Primordial  $\text{N}_2$  from the deep crust or mantle can migrate along the active deep-setting faults and accumulate (Krooss et al., 1995). The  $^{15}\text{N}$  values of primordial  $\text{N}_2$  should be similar to that of  $\text{N}_2$  in the atmosphere, because atmospheric  $\text{N}_2$  is mainly from the upper mantle degassing during the earth early evolution

(Zhang and Zindler,1989). Primordial  $^{15}\text{N}$  values of  $\text{N}_2$  are supposed to vary from  $-2\text{‰}$  to  $+4\text{‰}$  by Prasolov et al.(1990). The  $^{15}\text{N}$  values of molecular nitrogen derived from reduction of nitrogen-bearing salt minerals in evaporite are typically equal to  $+4\text{‰}$ (Prasolov et al.,1987).Littke et al.(1995) and Krooss et al.(1995) inferred that sedimentary organic matter and ammonium-clay minerals are the main sources of  $\text{N}_2$  in gas accumulations

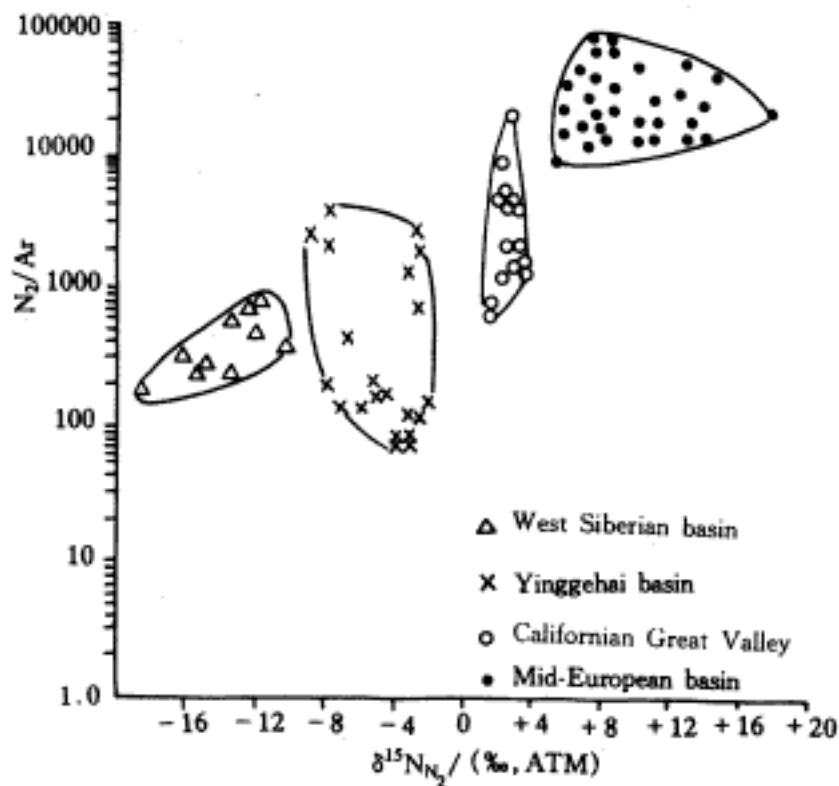


Fig.4 Semilogarithmic plot of  $\text{N}_2/\text{Ar}$  ratios versus  $\delta^{15}\text{N}_{\text{N}_2}$  of natural gases from the West Siberian (Prasolov et al.,1990), Yinggehai basin, Californian Great Valley (Jenden et al., 1988a) and Mid-European basin (Boigk et al.,1976; Stahl,1977;Weinlich 1991)

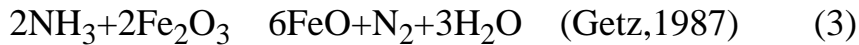
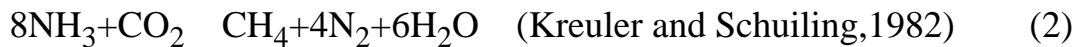
### $\text{N}_2$ from immature sedimentary organic matter

The immature sedimentary organic matter contains proteins and amino-acids in high concentration. Proteins can easily form amino-acids by hydrolysis, such as  $\text{COOH-CH}_2\text{-NH}_2$ , and amino-acids are unstable and ammoniated by microorganism:



The  $\text{NH}_3$  in this reaction is unstable. Part of it is absorbed by clay minerals, such as illite. Another part can transform into molecular nitrogen through the following reactions:





The  $\text{N}_2$  from immature sedimentary organic matter is depleted in heavy nitrogen (Macko et al., 1987).

The molecular nitrogen of gas pools in the Upper Cretaceous of West-Siberian basin is a typical one from immature sedimentary organic matter.  $^3\text{He}/^4\text{He}$  ratios of the associated helium ( $10^{-8}$ ) suggest that no significant  $\text{N}_2$  from the atmosphere or the deep crust or mantle is present. The  $^{15}\text{N}$  values do not indicate a contribution from nitrogen-bearing salt minerals in evaporite (Prasolov et al., 1990).

Thus the molecular nitrogen in the Upper Cretaceous of the West-Siberian basin should be derived from immature sedimentary organic matter, with typical  $^{15}\text{N}$  values varying from  $-19\text{‰}$  to  $-10\text{‰}$ .  $\text{N}_2$  with similar  $^{15}\text{N}$  values has also been discovered in the Quaternary of the Qaidamu basin, China. The  $^{15}\text{N}$  values of  $\text{N}_2$  from immature sedimentary organic matter should vary from  $-19\text{‰}$  to  $-10\text{‰}$ .

### **$\text{N}_2$ from mature sedimentary organic matter**

Although part of sedimentary organic nitrogen has been depleted during the immature stage, quite some organic nitrogen still exists in mature sedimentary organic matter, such as protein, amino-acid, pyridine and pyrrole, etc (Baxby et al., 1994), which is the reason why the nitrogen content in sedimentary rocks is the highest in the crustal rocks. During the mature stage, especially in the high mature stage, temperatures are high and approach those necessary to reach the activation energy (40 to 55 kcal/mol) to break nitrogen bonds in mature kerogen (Krooss et al., 1995).

Except for a small part of  $\text{NH}_3$  generated during this stage which is absorbed by authigenic clay minerals such as illite and montmorillonite in sedimentary rocks (Cooper and Erans, 1983; Oh et al., 1988), large volumes molecular of nitrogen are also generated through reaction (2) and (3) mentioned in section 4.1. These transformations are the possible reasons of  $\text{N}_2$ -rich gas pools associated with "red beds" and the low content of  $\text{CO}_2$  in nature. Via the  $\text{NH}_3$  stage, the generated molecular nitrogen is also depleted in heavy nitrogen ( $^{15}\text{N}$ ), but is heavier than that from immature sedimentary organic matter. In general, the molecules with the light isotope would decompose more readily, enriching  $^{14}\text{N}$  in the gas phase relative to the remaining nitrogen-bearing organic matter (Hoering and Moore, 1958).

The natural gases in the Pliocene Yinggehai formation of Yinggehai basin have the basic feature of an origin from mature (including high mature) sedimentary organic matter.  $^3\text{He}/^4\text{He}$  ratios of the associated helium vary from  $5.0 \times 10^{-8}$  to  $6.79 \times 10^{-7}$ , which suggests that there is little contribution of mantle-derived gas. No trace of magmatism or

nitrogen-bearing salt minerals has been discovered up to now in the basin.

mature (including high mature) sedimentary organic matter, with a typical isotopic feature of  $^{15}\text{N}$  values varying from -9‰ to -2‰. The  $\text{N}_2$  with this origin has been discovered in the Lower Cretaceous of the West Siberian basin (Prasolov et al., 1990) and Western Interior basin (Hoering and Moore, 1958; Jenden et al., 1988b). Thus, the range of  $^{15}\text{N}$  values of  $\text{N}_2$  from mature organic matter should vary from -10‰ to -2‰.

### **$\text{N}_2$ from post-mature sedimentary organic matter**

The geothermal energy at the mature stage of sedimentary organic matter is not high enough to release all organic nitrogen. Therefore organic nitrogen still exists in postmature sedimentary organic matter in high concentration. During the postmature stage, the geothermal energy is high enough to generate molecular nitrogen through pyrolysis (Krooss et al., 1995). Thus, a large amount of molecular nitrogen can be generated during this stage.

Littke et al. (1995) and Krooss et al. (1995) investigated the  $\text{N}_2$  generation from post-mature sedimentary organic matter at high temperature by laboratory pyrolysis experiments. They discovered that the peak of  $\text{N}_2$ -high gas generation was after that of methane-rich gas generation, and using the chemical dynamics method, they concluded that the essential geotemperature to generate  $\text{N}_2$ -high gas from organic matter should be at least 300 °C in nature.

Natural gases reservoirs in Permian Rotliegend and Triassic Buntsandstein formations of the Mid-European basin, especially in its northern part, have typical features of the generation from post-mature sedimentary organic matter. Marine evaporates were deposited in this basin, but there are no saltpeter or other nitrogen-bearing salt minerals in significant amount (Littke et al., 1995). Therefore the hypothesis of evaporite-derived  $\text{N}_2$  (Prasolov et al., 1990) has been denied. Although the magmatism is obvious, its contribution to gas accumulation can be neglected. On the contrary, the magma heat should have had accelerated the  $\text{N}_2$  generation from sedimentary organic matter at high temperature (Krooss et al., 1995).

Most of the molecular nitrogen, in Permian Rotliegend and Triassic Buntsandstein formations, should be generated from the Upper Carboniferous post-mature source rocks. The typical feature of molecular nitrogen  $^{15}\text{N}$  values from this type of generation should range from +4‰ to +18‰. The occurrence of  $\text{N}_2$  with this origin in a large scale may indicate that it is mainly the post-mature gas reservoirs in petroliferous basins (Littke et al., 1995; Krooss et al., 1995). Heavy  $\text{N}_2$  ( $^{15}\text{N}$  +5‰ ~ +18‰) discovered in the rift basin systems of eastern China, American Arkansas and Volga-Ural basin may be of the same origin.

### **$\text{N}_2$ from ammonium clay minerals**

The inorganic fixed nitrogen ( $\text{NH}_4^+$ ) in sedimentary rocks is primarily derived from organic matter. Some part of  $\text{NH}_3$  from immature and mature sedimentary organic matter

can be absorbed by clay minerals generating during the diagenesis of sedimentary rocks. In

clay minerals (Cooper and Erans, 1983; Oh et al., 1988; Baxby et al., 1994; Littke et al., 1995; Krooss et al., 1995). The fixed nitrogen in clay minerals shows a high thermal stability and cannot release molecular nitrogen unless at a high temperature. This temperature should be higher than 1000 °C during laboratory pyrolysis experiment and higher than 500 °C under geological conditions according to chemical dynamic calculation (Whelan et al., 1988). Thus, molecular nitrogen can be generated from metamorphism of ammonium clay mineral-bearing sedimentary rocks, with the  $^{15}\text{N}$  value varying from +1‰ to +4‰.

The natural gases in the Cretaceous and Tertiary of the Californian Great Valley have complex origins involving mixing of multiple sources, with a narrow range of molecular nitrogen  $^{15}\text{N}$  values from +0.9‰ to +3.5‰. Jenden et al. (1988a) suggested this kind of molecular nitrogen should originate from Upper Jurassic Franciscan metasedimentary complex. The isotopic composition of  $\text{N}_2$  from ammonium clay mineral may vary from +1‰ to +4‰.

## CONCLUSIONS

There are multiple origins of  $\text{N}_2$  in the subsurface and they often mix together in  $\text{N}_2$ -bearing gas reservoirs in petroliferous basins. Therefore widely ranging  $^{15}\text{N}$  values of  $\text{N}_2$  have been observed and are puzzling for scientists. Based on analyzing and contrasting four typical  $\text{N}_2$ -bearing basins, West-Siberian basin, Yinggehai basin, Mid-European basin and Great Valley, and the study of the molecular nitrogen possible origins and the corresponding isotopic composition features, we can conclude on the isotope geochemistry of molecular nitrogen. Values between -19‰  $^{15}\text{N}_{\text{N}_2}$  -10‰ represents the  $\text{N}_2$  from immature sedimentary organic matter.  $\text{N}_2$  with -10‰ <  $^{15}\text{N}_{\text{N}_2}$  -2‰ originates from mature (including high-mature) sedimentary organic matter. -2‰ <  $^{15}\text{N}_{\text{N}_2}$  < +1‰ may reflect the  $\text{N}_2$  from the mantle.  $^{15}\text{N}_{\text{N}_2}=0‰$  and  $\text{N}_2/\text{Ar}=38 \sim 84$  characterizes  $\text{N}_2$  from atmosphere. +1‰  $^{15}\text{N}_{\text{N}_2}$  < +4‰ indicates  $\text{N}_2$  metamorphism of ammonium clay mineral-bearing sedimentary rocks.  $^{15}\text{N}_{\text{N}_2}=4‰$  is the isotopic composition feature of  $\text{N}_2$  from saltpeter in evaporite. Values between +4‰ <  $^{15}\text{N}_{\text{N}_2}$  +18‰ imply that  $\text{N}_2$  may originate from post-mature sedimentary organic matter. The isotopic compositions of  $\text{N}_2$  from multiple origins appear irregular.

Gas pools containing  $\text{N}_2$ -high ( $\text{N}_2 > 60\%$ ) and  $\text{N}_2$ -rich ( $\text{N}_2 > 15\%$ ) may be dominantly from post-mature sedimentary organic matter. The occurrence of  $\text{N}_2$ -high gas pools in a large scale may indicate that this gas is mainly derived from post-mature source rocks in

petroliferous basins.  $N_2$ -rich gas ( $N_2 > 15\%$ ) may also originate from mature, especially high

be formed by metamorphism of ammonium clay mineral-bearing sedimentary rocks.

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